

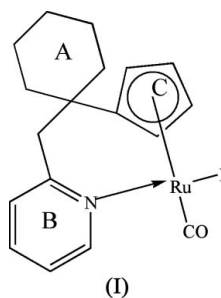
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## Key indicators

Single-crystal X-ray study  
 $T = 296$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.008$  Å  
 $R$  factor = 0.036  
 $wR$  factor = 0.119  
Data-to-parameter ratio = 20.1For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.Carbonyliodo{[1-(2-pyridylmethyl)cyclohexyl-  
methyl]- $\eta^5$ -cyclopentadienyl- $\kappa N$ }ruthenium(II)In the title compound,  $[\text{Ru}(\text{C}_{17}\text{H}_{20}\text{N})\text{I}(\text{CO})]$ , the distance  
between the Ru atom and the centroid of the cyclopentadiene  
ring is 1.850 (2) Å. The dihedral angle between the planes of  
the pyridine and cyclopentadiene rings is 53.8 (2)°.

## Comment

The chemistry of cyclopentadiene metal complexes containing  
a donor-functionalized side chain has been receiving extensive  
attention (Jutzi & Redeker, 1998; Jutzi & Siemeling, 1995). For  
the pyridyl side-chain-functionalized cyclopentadiene ligands,  
the N atom can act as a good two-electron donor site and can  
coordinate to Ru, such as in the title compound, (I).

In (I) (Fig. 1), the Rh—Cp\* distance (Cp\* is the centroid of the cyclopentadiene ring) is 1.850 (2) Å. The Cp\*—Ru1—I, Cp\*—Ru1—N1 and Cp\*—Ru1—C18 angles are 123.75 (10), 122.80 (12) and 128.29 (17)°, respectively. The dihedral angle between the planes of the pyridine and cyclopentadiene rings is 53.8 (2)°. The cyclohexane ring adopts a chair conformation. In the crystal structure, inversion-related molecules form a C—H···O hydrogen-bonded dimer (Table 2).

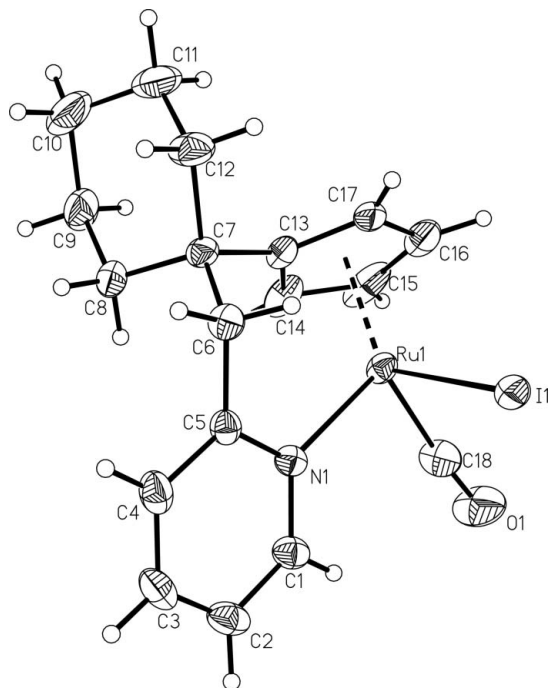
## Experimental

Bis([1-(pyridin-2-ylmethyl)cyclohexyl]methyl)cyclopentadienyl-tetra(carbonyl) diruthenium, (II) (0.705 mmol), was prepared from 2-[1-(cyclopenta-2,4-dienyl)cyclohexyl]pyridine (1.41 mmol) and dodecacarbonyltriruthenium (0.46 mmol) under refluxing xylene for 12 h (Chen *et al.*, 2006). Compound (I) was prepared from compound (II) (0.705 mmol) and iodine (0.705 mmol). Red single crystals suitable for X-ray diffraction were obtained by slow evaporation of a dichloromethane–hexane solution (1:1) at room temperature.

## Crystal data

$[\text{Ru}(\text{C}_{17}\text{H}_{20}\text{N})\text{I}(\text{CO})]$   
 $M_r = 494.32$   
Monoclinic,  $P2_1/c$   
 $a = 6.7349$  (13) Å  
 $b = 22.903$  (5) Å  
 $c = 11.626$  (2) Å  
 $\beta = 99.29$  (3)°  
 $V = 1769.8$  (6) Å<sup>3</sup>

$Z = 4$   
 $D_x = 1.855$  Mg m<sup>-3</sup>  
Mo  $K\alpha$  radiation  
 $\mu = 2.63$  mm<sup>-1</sup>  
 $T = 296$  (2) K  
Block, red  
0.36 × 0.32 × 0.30 mm



**Figure 1**  
A view of the molecular structure of (I). Displacement ellipsoids are drawn at the 30% probability level.

**Data collection**

Rigaku R-AXIS RAPID IP area-detector diffractometer	17030 measured reflections
$\omega$ scans	4026 independent reflections
Absorption correction: multi-scan (ABSCOR; Higashi, 1995)	3435 reflections with $I > 2\sigma(I)$
$T_{\min} = 0.451$ , $T_{\max} = 0.506$ (expected range = 0.405–0.454)	$R_{\text{int}} = 0.057$ $\theta_{\text{max}} = 27.5^\circ$

**Refinement**

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.063P)^2 + 0.677P]$
$R[F^2 > 2\sigma(F^2)] = 0.036$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.119$	$(\Delta/\sigma)_{\text{max}} = 0.002$
$S = 1.13$	$\Delta\rho_{\text{max}} = 0.95 \text{ e } \text{Å}^{-3}$
4026 reflections	$\Delta\rho_{\text{min}} = -1.37 \text{ e } \text{Å}^{-3}$
200 parameters	Extinction correction: SHELXL97
H-atom parameters constrained	Extinction coefficient: 0.0093 (8)

**Table 1**  
Selected geometric parameters (Å, °).

N1—Ru1	2.139 (3)	Ru1—C16	2.200 (5)
Ru1—C18	1.872 (5)	Ru1—C17	2.242 (5)
Ru1—C14	2.173 (5)	Ru1—C13	2.245 (4)
Ru1—C15	2.178 (5)	Ru1—I1	2.7016 (8)
C18—Ru1—N1	95.00 (18)	N1—Ru1—I1	87.17 (9)
C18—Ru1—I1	88.43 (19)		

**Table 2**  
Hydrogen-bond geometry (Å, °).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
C15—H15 $\cdots$ O1 <sup>i</sup>	0.98	2.53	3.405 (9)	148

Symmetry code: (i)  $-x, -y + 1, -z + 1$ .

H atoms were placed in calculated positions and constrained to ride on their parent atoms, with C—H distances of 0.93–0.98 Å and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ . In the absence of significant anomalous scattering effects, Friedel pairs were averaged. The deepest hole in the difference map is located at (1.0283, 0.1801, 0.4920).

Data collection: RAPID-AUTO (Rigaku/MSC, 2004); cell refinement: RAPID-AUTO; data reduction: RAPID-AUTO; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Bruker, 2000); software used to prepare material for publication: SHELXL97.

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